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2426 Electron = -1.602×10^{-19} C = 9.11×10^{-31} kg Proton = 1.602×10^{-19} C = 1.67×10^{-27} kg Neutron = 1.67×10^{-27} kg 6.022×10^{23} atoms in one atomic mass unit e is the

elementary charge: 1.602×10^{-19} C Potential Energy, velocity of electron: $PE = eV = \frac{1}{2}mv^2$ $1V = 1J/C$ $1N/C = 1V/m$ $1J = 1 N \cdot m = 1 C \cdot V$

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